

Article

# Preparation, Characterization and Evaluation of Biological Activity and Study of Molecular Docking of Some New Thiazolidine Derivatives

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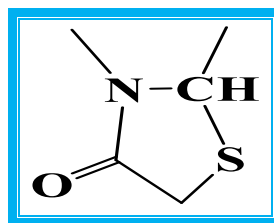
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**Abstract:** In this work, equal moles of hydrazone base derivatives and thioglycolic acid react to form pentacyclic thiazolidine derivatives with zinc chloride as a catalyst and using dioxane as a solvent. Proton nuclear magnetic resonance spectroscopy, infrared spectroscopy, and physical and spectroscopic techniques have confirmed the complex structures. Furthermore, the purity and melting points were determined, and thin-layer chromatography (TLC) was used to track the reaction development. The researchers examined how several chemicals produced affected the development of two bacterial isolates: Escherichia coli, gram-negative, and Staphylococcus aureus, gram-positive. Using MOE software (2009), molecular docking probes of compounds (AB20, AB21) against E. coli were performed. The most stable form of these molecules (lowest energy barrier) was achieved using an energy minimization procedure.

**Keywords:** hydrazone, thiazolidinone, biological activity, molecular docking

## 1. Introduction

One class of derivatives of thiazolidines is thiazolidinone compounds. These are heterocyclic compounds with nitrogen and sulfur; they are called thiazolidine-4-one when they have a carbonyl group at position -4. Compared to thiophene, which is stable against acid at moderate temperatures, these systems seem more stable [1].



Thiazolidinedione compounds are biologically active. It is essential because it contains a sulfur and nitrogen atom. It has developed a series of antidiabetic agents for type 2 diabetes and mainly studied the blood sugar level and the extent to which the cell genetically reduces its activities in resisting obesity. This means that cells sometimes become resistant to insulin, as it has caused a significant breakthrough in Antidiabetic treatment by increasing insulin secretion, so it is also called "insulin sensitizers" [2]. Both compounds are used to treat diarrhoea [3]. It also showed high effectiveness as an anti-cancer [4], anti-bacterial [5], anti-fungal agent [6], in addition to being an anti-inflammatory [7].

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## 2. Materials and Methods

### 2.1. Material

Without any additional purification, all of the compounds utilized in this investigation were acquired from BDH, Fluka, and Aldrich.

### 2.2. Devices used

A thermoelectric melter 9300 was used to determine melting points. Using KBr disk at a scale of (400–4000)  $\text{cm}^{-1}$ , Shimadzu FT-IR 8400S spectrophotometer;  $^1\text{H}$ -NMR and  $^{13}\text{C}$ -NMR spectra using Bruker apparatus operating at 400 MHz. Thicknessed at 0.2 mm, Fluka silica gel plates were employed in thin-layer chromatography (TLC). The plates were activated with fluorescent silica gel G, and visibility was achieved by UV light.

### 2.3. Preparation of thiazolidinone [8], [9]

A mixture was raised in 40 milliliters of dry gasoline in a water bath with continuous stirring. The ingredients were prepared hydrazone derivatives [AB2-AB6] (0.01 mol), anhydrous zinc chloride (0.01 mol), and thioglycolic acid (0.01 mol, 0.7 g). After confirming that the reaction had finished using thin-layer chromatography, the solution was left to cool for seven hours. Remote heating was used to extract the solvent during this time. It was filtered and recrystallized with ethanol to give a precipitate, and Table 1 shows some physical properties of thiazolidinone derivatives [AB17-AB21].

### 2.4. Evaluation of biological activity [10]

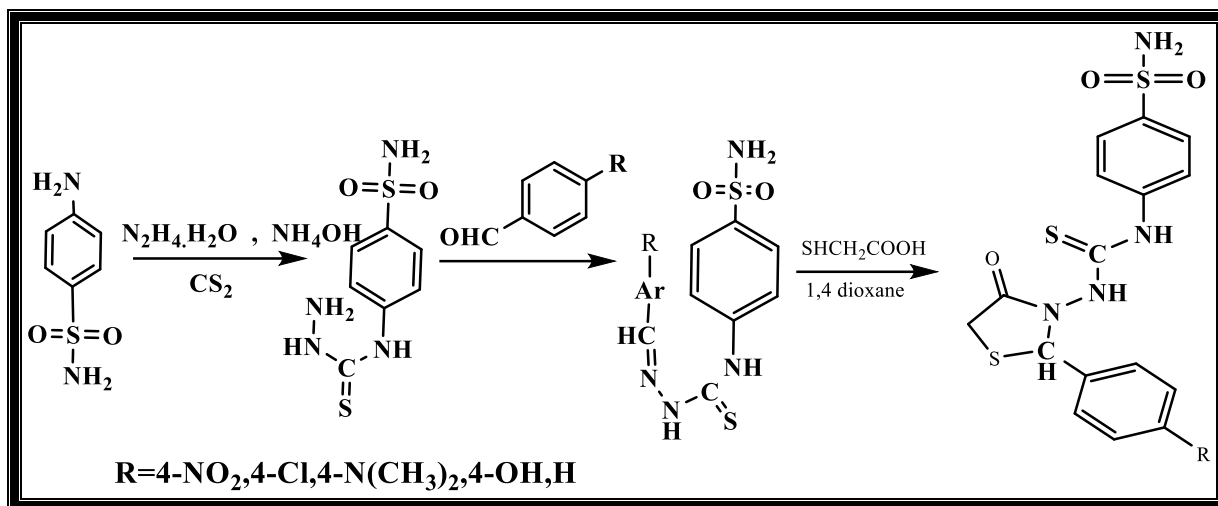
Bacteria species stock cultures were overnight activated in nutrient broth; inoculum was prepared by diluting activated bacteria with 0.8 % normal saline to obtain cell concentration of about  $1.5 \times 10^8$  cell/ml, the turbidity of bacteria solution adjusted visually to 0.5 McFarland standard via Twickenham card then the obtained inoculum was streaked directly on agar plates (MHA) by using sterile cotton swabs, wells (holes) with a diameter of 6 mm was made subsequently on the surface of these plates where the solution will be loaded [11]. The negative control plates were maintained with autoclaved plate loaded with DMSO. In contrast, positive control was standard antibiotic disk [12].

### 2.5. Molecular docking study of some prepared compounds [13], [14]

Using the MOE program (2009), molecular docking investigations were carried out for a few produced compounds (AB12, AB13) against a common bacterial strain, *Escherichia coli*. The goal was to reduce the energy of the compounds under study to achieve the lowest energy barrier or the most stable conformation. The *Escherichia coli* protein structure was obtained from the International Protein Bank. High-performance computing resources were utilized because these programs are very demanding and require sophisticated, multi-core processors for quick and effective computational operations, especially when working with giant molecules and complex atom configurations, among other things.

## 3. Results and Discussion

This study included the preparation of thiazolidinone derivatives belonging to the class of pentacyclic rings from the reaction of hydrazones prepared with thioglycolic acid in the presence of zinc chloride as a catalyst, using dioxane as a solvent, as in Scheme 1.



**Scheme 1.** Path of the ready compounds (AB17-AB21)

### 3.1. Characterization of thiazolidinone

Absorption bands at the D range (1243-1281) and in the range (1685-1701)  $\text{cm}^{-1}$  were identified in the infrared spectrum and were attributed to the thiazolidines' (C=O) group. Two bands at (1456-1506)  $\text{cm}^{-1}$  and (1568-1595)  $\text{cm}^{-1}$  are attributed to the group (C=C), bands at (3035-3078)  $\text{cm}^{-1}$  are attributed to (Ar-C-H), and absorption bands at (2928-2988)  $\text{cm}^{-1}$  are attributed to (C-H) affinity [15]. The absorption bands are caused by the (C-N) group in the same ring as in Table 2 and Figure 1 and 2.

Upon analyzing the  $^1\text{H-NMR}$  spectrum, several signals were detected within the 7.16–7.64 ppm range, and a signal associated with the protons of the (NH) group was detected at the 8.24–10.94 ppm range. The proton (CH<sub>2</sub>) of the thiazolidine ring was found to be the source of a signal at position (3.94) ppm, while the protonation of the thiazolidine ring connected to the aromatic ring (CH) was responsible for a signal at location (5.11) ppm [16] as in Figure 3.

The carbon of the (C=O) group in the thiazolidine ring was found to be the source of a signal at (165.79) ppm in the  $^{13}\text{C-NMR}$  spectrum. The carbon of the (CH<sub>2</sub>) group was identified as the source of a signal at (18.24) ppm, and the carbon of the (CH) group was identified as the source at (59.66) ppm. It has a multiple signal at (122.99-152.65) ppm attributed to the aromatic ring's carbons in addition to the same ring linked to the aromatic ring [17] as in Figure 4.

### 3.2. Evaluation of biological activity of (AB17, AB18, AB19, AB20)

Some generated compounds (AB17, AB18, AB19, and AB20) were tested against various bacterial strains using the cup plate agar diffusion technique, including gram-positive bacteria *Staph. Aureus*, as well as gram-negative *E. Coli* [18] 0.8% sterile saline was added to the microbial cultures after incubating them for eight hours at 37 °C [19]. The concentration of the drug solution in DMSO was maintained at 100  $\mu\text{g}/\text{mL}$ . Ciprofloxacin was employed as a negative control. The biological activity was determined by measuring the inhibition diameter of bacterial growth surrounding the in-use disk [20], [21] as seen in Table 3.

### 3.3. Results of molecular docking study for some prepared compounds [22]

The molecular docking of some of the prepared compounds (AB20, AB21) was studied on one line: *Pseudomonas aeruginosa* bacteria. The values of the binding energies of the prepared compounds were calculated using the MOE program (2009). As shown in Table

4, the study showed that the compound (AB20) interacts with amino acid residues, which are present in the active site, by forming one type of bond, three hydrogen bonds. The first connects the residue of the amino acid CYS 1 present in the active site with the electronic pair of the oxygen atom of the hydroxyl group, and two hydrogen bonds connect the amino acid residues ASP192, GLN193 is located in the enzyme's active site with the electron pair of the sulfur atom, the thiocarbamide group. When studying the compound (AB21), a single hydrogen bond was formed linking the amino acid residue GLN193 located in the active site with the electronic pair of the oxygen atom of the carbonyl group in the thiazolidine ring as in Figure 5 and 6.

**Table 1.** The produced chemicals' physical attributes (AB17-AB21)

Comp. No.	R	Molecular formula	m.p. °C	Yield%	Color
AB17	4-NO <sub>2</sub>	C <sub>16</sub> H <sub>15</sub> N <sub>5</sub> O <sub>5</sub> S <sub>3</sub>	191-193	65	White
AB18	4-Cl	C <sub>16</sub> H <sub>15</sub> ClN <sub>4</sub> O <sub>3</sub> S <sub>3</sub>	216-218	71	Brown
AB19	4-N(CH <sub>3</sub> ) <sub>2</sub>	C <sub>18</sub> H <sub>21</sub> N <sub>5</sub> O <sub>3</sub> S <sub>3</sub>	286-288	63	Blue
AB20	4-OH	C <sub>16</sub> H <sub>16</sub> N <sub>4</sub> O <sub>4</sub> S <sub>3</sub>	223-225	81	Yellow
AB21	4-H	C <sub>16</sub> H <sub>16</sub> N <sub>4</sub> O <sub>3</sub> S <sub>3</sub>	254-256	69	Orange

**Table 2.** The synthesized compounds' FT-IR data (AB17-AB21) cm<sup>-1</sup>

Comp. No.	R	IR (KBr) cm <sup>-1</sup>						
		v(C-H) Arom.	v(C-H) Aliph.	v C=O)	v(N-H)	v(C=C) Arom.	v(C=N)	Others
AB17	4-NO <sub>2</sub>	3053	2960	1699	3174	1467,1595	1276	v (N-O) 1332
AB18	4-Cl	3078	2967	1685	3216	1447,1557	1254	v (C-Cl) 741
AB19	4- N(CH <sub>3</sub> ) <sub>2</sub>	3035	2925	1701	3180	1506,1591	1244	--
AB20	4-OH	3037	2988	1691	3165	1481,1568	1281	v (OH)3381
AB21	4-H	3058	2628	1687	3171	1456,1592	1243	---

**Table 3.** Antibiotics and other generated chemicals (AB17, AB18, AB19, AB20) have the ability to prevent the growth

Comp. No.	E. Coil Conc. mg/ml			Staph. Aureus Conc. mg/ml		
	0.01	0.001	0.0001	0.01	0.001	0.0001
AB17	18	14	10	15	10	10
AB18	15	15	10	10	5	--
AB19	10	5	5	5	5	-
AB20	15	12	8	18	15	10
Amoxicillin	12	10	10	21	20	10

**Table 4.** Values of the produced compounds' binding energies

Comp. No.	RMSD	Docking Score
AB20	2.24	-6.51
AB21	2.08	-7.074

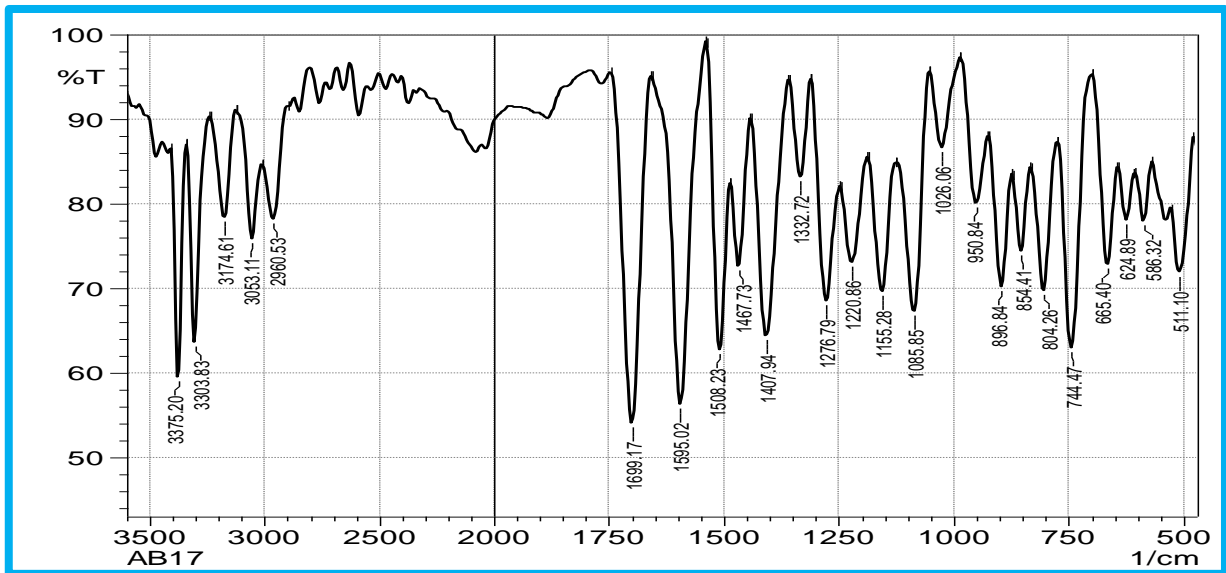


Figure 1. The compound's infrared spectrum (AB17)

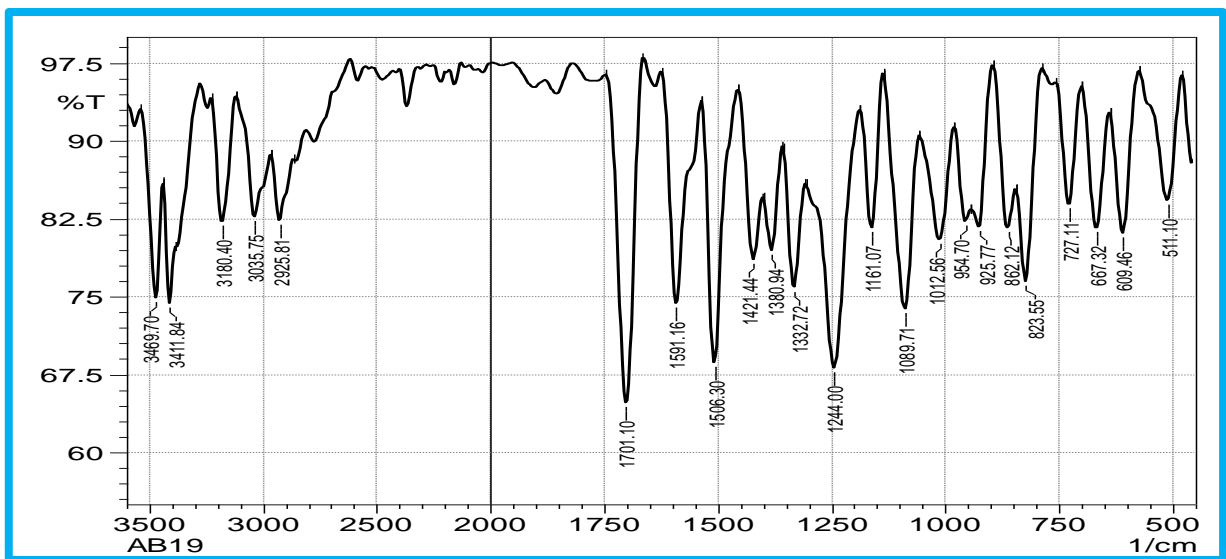


Figure 2. The compound's infrared spectrum (AB19)

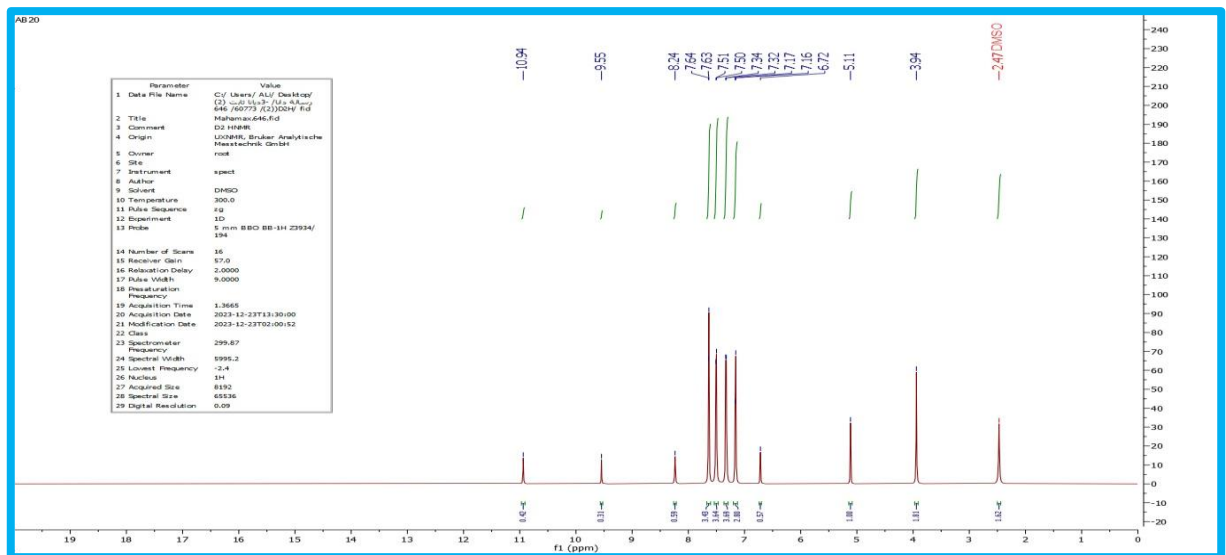


Figure 3. The chemical's <sup>1</sup>H-NMR spectra (AB20)

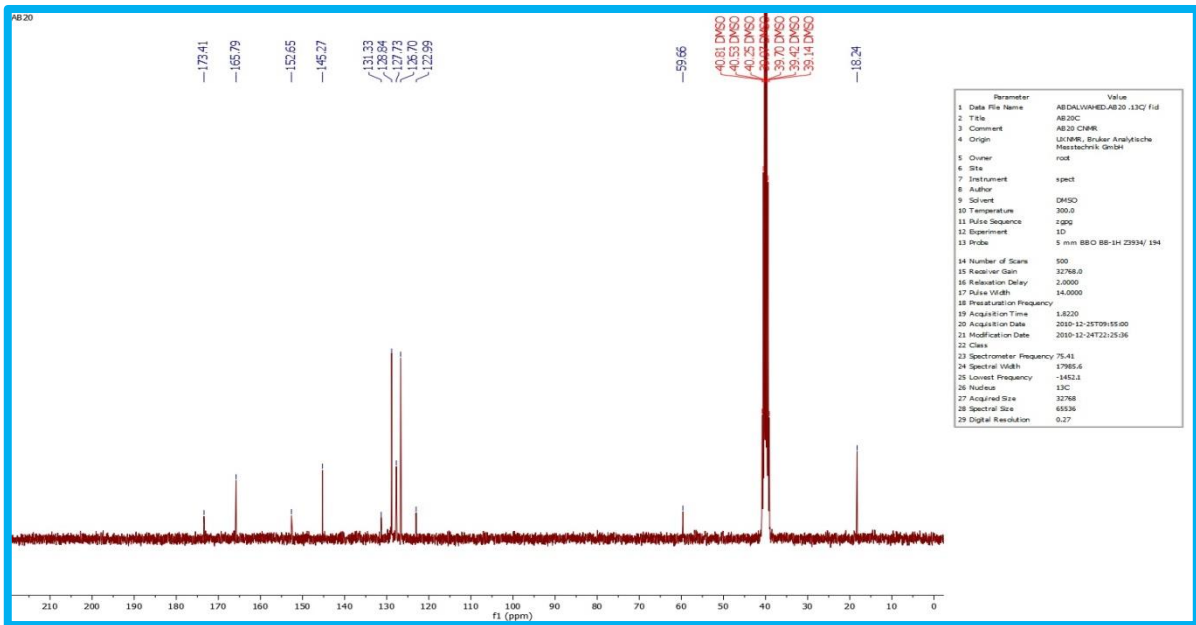


Figure 4. <sup>13</sup>C-NMR spectra of the substance (AB20)

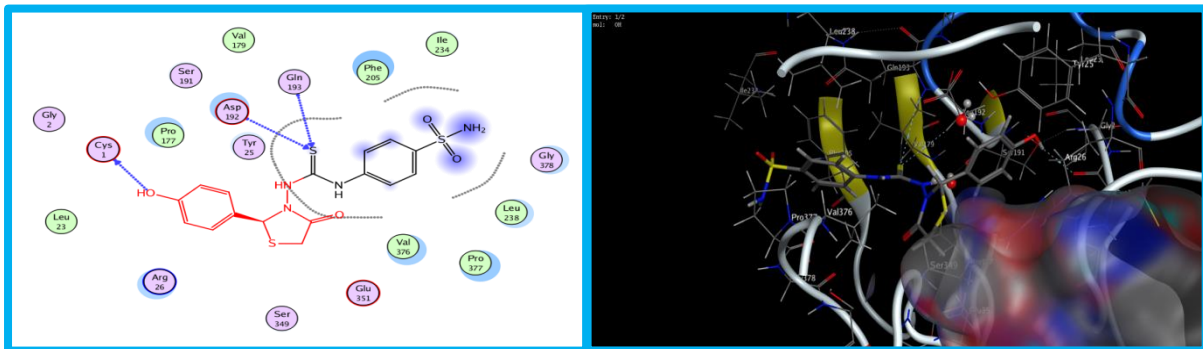


Figure 5. Compound (AB20) interactions in two and three dimensions

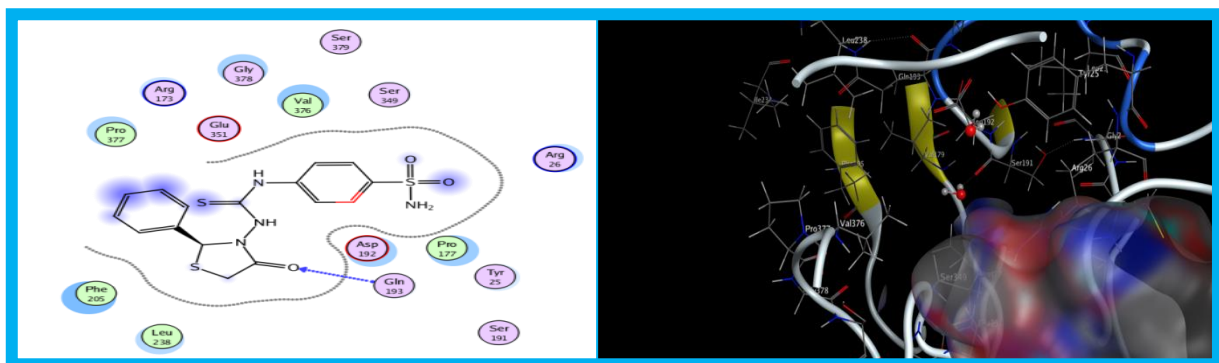


Figure 6. Compound (AB21) interactions in two and three dimensions

#### 4. Conclusion

The reaction of hydrazone derivatives with compounds containing appropriate functional groups often gives rise to heterocyclic pentacyclic rings. Bioanalysis results indicate that most prepared compounds showed antibacterial activity and could inhibit bacterial growth. Some of these compounds showed higher biological activity than the antibiotics used as control samples. Physical and spectroscopic measurements showed the prepared compounds' composition accuracy, as the measurements showed that the prepared compounds were of high purity.

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